



# Atomic Layer Deposition of Palladium Films on $\text{Al}_2\text{O}_3$ Surfaces

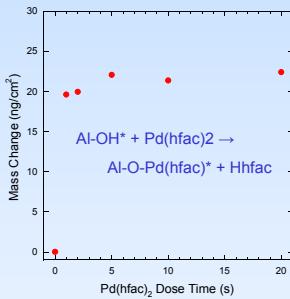
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## Abstract

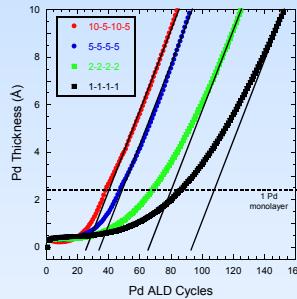
ALD Pd films were deposited at 200 °C on ALD  $\text{Al}_2\text{O}_3$  surfaces using sequential exposures to Pd(II) hexafluoroacetylacetone (Pd(hfac)<sub>2</sub>) and formaldehyde. This technique allows Pd films to be deposited with atomic layer precision on any surface by first depositing a thin ALD  $\text{Al}_2\text{O}_3$  seed layer. *In situ* quartz crystal microbalance measurements revealed that the Pd nucleation is hampered by the relatively slow reduction of the adsorbed Pd(hfac)<sub>2</sub> species, but is accelerated using larger initial reactant exposures. Pd films were deposited on silicon, glass and mesoporous anodic alumina (AAO) following the ALD of a 1-2 nm  $\text{Al}_2\text{O}_3$  seed layer. These samples yielded a Pd ALD growth rate of 0.2 Å/cycle following a nucleation period of slower growth. The films are pure, cubic phase Pd and are relatively rough and highly conductive. ALD Pd films deposited on the inside walls of very high aspect ratio AAO membranes show promise for hydrogen sensors.

## Pd Nucleation on $\text{Al}_2\text{O}_3$ using QCM

- Many potential reducing and oxidizing agents were screened ( $\text{H}_2$ , alcohols, acetone, TMA,  $\text{O}_3$ ,  $\text{H}_2\text{O}_2$  etc). Only formaldehyde (HCOH) enabled nucleation of Pd on  $\text{Al}_2\text{O}_3$ .

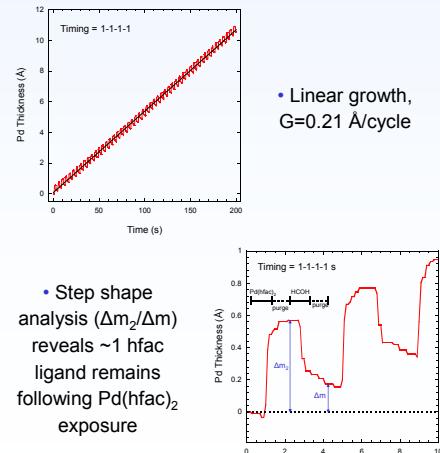
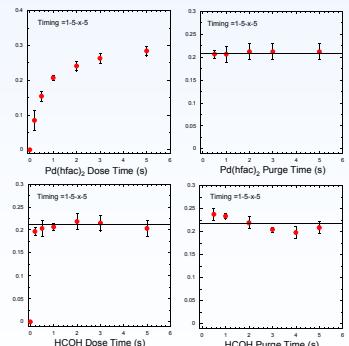


- Pd(hfac)<sub>2</sub> reaction saturates rapidly, but reduction after first exposure is slow



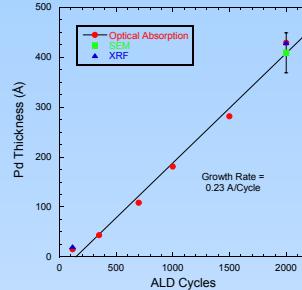
- Linear growth,  $G=0.21 \text{ \AA}/\text{cycle}$

## Growth of Pd using QCM

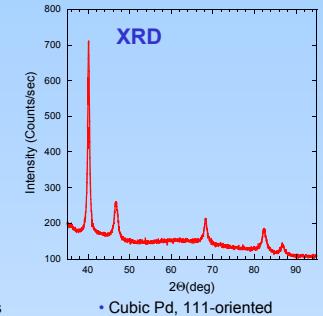


## Growth and Analysis of Pd Films

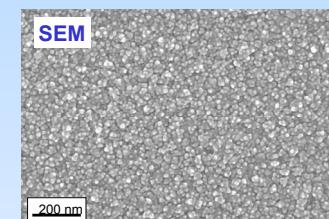
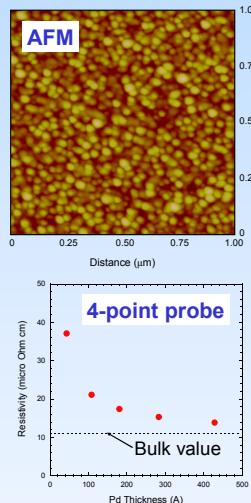
- ALD Pd films deposited on glass and Si(111) following 2-nm  $\text{Al}_2\text{O}_3$  buffer layer



- Nucleation and growth confirm QCM results



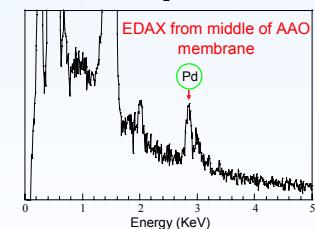
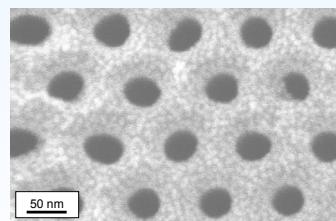
- Cubic Pd, 111-oriented



- AFM shows nanocrystalline surface, RMS roughness=4.2 nm for 42 nm thick film
- SEM shows nanocrystalline surface
- 4-point probe resistivity=14  $\mu\Omega \text{ cm}$  for 42 nm thick film

## Pd ALD in AAO for Hydrogen Sensing

- AAO membrane: pore diameter  $d=40 \text{ nm}$ , pore length  $L=60 \mu\text{m}$ , aspect ratio  $L/d=1500$
- Coat AAO with ~2 nm ALD Pd, measure resistance change upon  $\text{H}_2$  exposure



- SEM shows nanocrystalline surface
- EDAX shows infiltration of Pd into center of AAO membrane
- Fast resistance increase during  $\text{H}_2$  exposures due to Pd-hydride formation

